The Study of Polymorphism in Pharmaceutical Operating Processes: Granulation Technique for Carbamazepine Tablets

Belsarkar AS*, Shinde AD, Adsul PS, Dhokale VB. SVPM's College of Pharmacy, Malegoan BK Department of Pharmaceutics

Abstract:- Epilepsy, trigeminal neuralgia, and bipolar disorder are just a few of the conditions that are treated with Carbamazepine, a Dibenzazepine derivative. First-line treatment for seizures that has exceptional efficacy and is inexpensive is Carbamazepine (CBZ). Due to their high membrane permeability and low water solubility, which make dissolution the rate-limiting step in the absorption process, they are categorised as BCS class II substances by Biopharmaceutical Classification polymorphic alterations of the drug Carbamazepine are examined in its tablet dosage form by taking into consideration variations in the granulation process or method. To discover the polymorphic alterations in Carbamazepine, a DSC study was conducted for a tablet formulation. This study is set out to investigation to identify the effects of various granulation techniques on the drug Carbamazepine for distinct polymorphic alterations in tablet dosage form.

Keywords:- Carbamazepine, Polymorphism, DSC, Variations, Epilepsy, Tablet.

I. INTRODUCTION

In Human Kind any age group can be impacted by epilepsy, a chronic brain condition. Around 50 million people worldwide suffer from epilepsy. Greek term epilepsy translates as "to be overpowered, seized, or besieged." The propensity for seizures that a person has is simply referred to as having "epilepsy." The brain is a fragile and intricate organ. This organ directs and regulates all of our behaviours, gestures, feelings, thoughts, and emotions. It regulates the body's automatic internal processes, including heart and lung function. Brain cells communicate with one another using electric signals. On rare occasions, a seizure could be brought on by an abnormal electrical discharge from a group of cells.

According to the BCS classification system, Carbamazepine is a class II medication. Currently administered orally, Carbamazepine has a low water solubility (170 g/ml), requiring high and frequent doses to maintain therapeutic effect. This is important because it affects the slow and erratic GI system's slow and variable medicine absorption, which causes variable Carbamazepine bioavailability. It can also affect a wide range of API properties such as tableting, dissolving rate, solubility, stability, and even biopharmaceutical performance such as safety and efficacy as well as toxicity, as well as the physicochemical features of these polymorphic forms of API varies/changes such as dissolution and solubility¹.

Polymorphism is important physicochemical properties, and it brings major polymorphic changes in drug,in unit operations, bulk manufacturing as well as in manufacturing dosage forms on a large scale.

II. EXPERIMENTAL

A. Material

The Carbamazepine pure drug was procured from Aarti Distributers, Mumbai and all other ingredients were procured Loba-Chem Pvt. Ltd.

B. Methods

a) Formulation and Development

The tablet's composition, which includes Carbamazepine, microcrystalline cellulose as a diluent, and magnesium stearate as a lubricant, was created to be hard and tough for the last dose. To improve medication compatibility and better understand the polymorphic changes that take place during the formation of Carbamazepine tablets, tablet formulation is created. Wet and Dry granulation was used to formulate the Carbamazepine tablet.

Sr. No	Name of Ingredients	Quantity taken for each tablet (Set Formula) (mg)	Quantity taken for each tablet (Set Formula) (gm)	% Strength (w/w)
1	Carbamazepine	100	0.1	33.33
2	Starch	4	0.004	1.33
3	Magnesium Stearate	1	0.001	65.33
4	Microcrystalline Cellulose /Lactose	Q.S 300	Q.S 0.3	Q.S 100

Table 1: Formulation of Carbamazepine Tablets

ISSN No:-2456-2165

b) Making Carbamazepine Tablets Through Dry and Wet Granulation:

According to the formulation table, the precise weights of all the constituents, including the medication and excipients, were recorded. After thoroughly combining all the materials, with the exception of the lubricant and disintegrant, the powder mixture was compressed into a slug. Following that, the slug was milled and sieved.

The next stage involved combining the granules with the disintegrant and lubricant before

compressing them into tablets. All materials, including the medication and excipients, were correctly weighed for wet granulation in accordance with the formulation table. The ingredients, with the exception of the lubricant, were all well combined after which the starch paste was added. The damp mass was then made, wet screened, and then the damp mass was screened into granules.

The wet granulation process using hydroxypropyl methyl cellulose as the polymer in place of the diluent microcrystalline cellulose.

C. Observations of Evaluations:

a) Pre-Compression Parameters of Carbamazepine Tablets:

Formulations	Angle of Repose	Loose of Bulk density (gm/ml)	Tapped Bulk Density (gm/ml)	Carr's Index	Hausner's Ratio
A1	19.98	0.588	0.683	13.85	1.11608
A2	19.71	0.596	0.685	12.98	1.1492

Table 2: Preformulation Parameters of Powder Mixture:

b) Flow Properties:

Determination of angle of repose, Carr's index and Hausner's ratio were used to characterize flow properties of powder which are used to prepare Conventional tablet of Carbamazepine by fixed funnel method. The flowability of a powder is of critical importance in the production of pharmaceutical dosage forms in order to get a

uniform feed as well as reproducible filling of tablet dies.

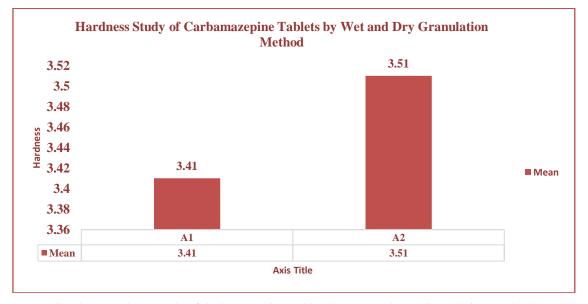
D. Post-Compresional Parameters of Carbamazepine Tablet:

a) Hardness:

Tablets should be sufficiently hard to resist breaking during normal handling. The hardness of tablets was measured by Monsanto hardness tester, hardness was measured in terms of kg/cm².

Formulations	Hardness (kg/cm²)								
	1	2	3	4	5	Mean	SD	SEM	
A1	3.2	3.2	3.7	3.1	3.1	3.41	±0.444	0.27	
A2	3.2	3.5	3.0	3.3	3.3	3.51	±1.026	0.41	

Table 3: Hardness of Carbamazepine Tablets



Graph 1: Hardness study of Carbamazepine Tablets by Wet and Dry Granulation Method.

ISSN No:-2456-2165

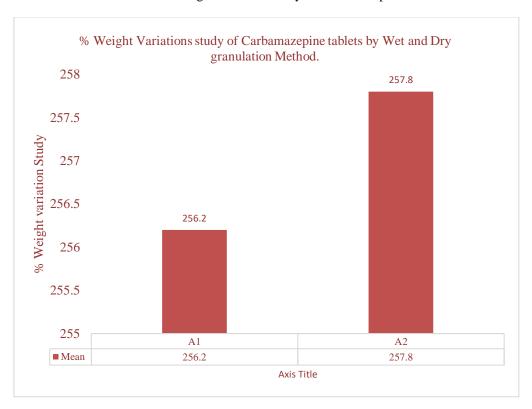
After compression, a tablet requires a certain amount of mechanical strength to withstand the shock of handling in the manufacture, packing, shipping and dispensing. The hardness of the tablet was found to be in the range of 3.41 to $3.55~{\rm kg/cm^2as}$ shown in table. This ensures that tablets with good mechanical strength

b) % Weight Variation:

The weight of tablet were measured to ensure that a tablet contain the proper amount of drug. Weight variation test was performed as per IP 2007. Twenty tablets were selected randomly and weighed. Average weight of the tablet was determined, not more than the two of the individual weights deviate from the average weight by more than 5% percentage deviation.

Formulation code	Mean	SD
A1	256.2	±1.356
A2	257.8	±2.324

Table No.4: % Weight Variation Study of Carbamazepine Tablet



Graph 2: % Weight variation of Study of Carbamazepine Tablets

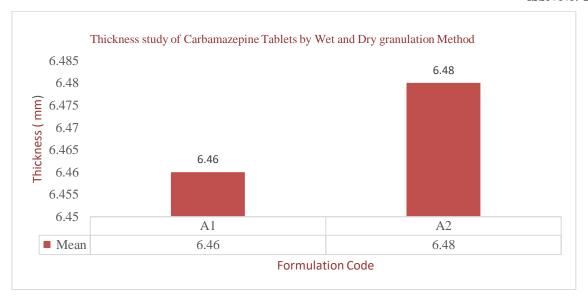
All the formulated A1 to A2 tablets shows 1.33%-2.36% weight variation was within the pharmacopoeial limits +5% of the weight. The weights of all the tablets were found to be uniform with low standard deviation values.

c) Thickness:

The six tablets from each batch of formulation were collected and the diameter of the tablets was measured with the help of electronic VernierCaliper. The average thickness was calculated.

Formulation		Thickness(mm)							
code	1	2	3	4	5	6	Mean	SD	SEM
A1	6.4	6.5	6.4	6.5	6.7	6.3	6.46	±0.136	0.055
A2	6.4	6.5	6.4	6.5	6.7	6.3	6.48	±0.147	0.060

Table 5: Thickness Study of Carbamazepine Tablets by Wet and Dry Granulation



Graph 3: Thickness Study of Carbamazepine Tablet by wet and Dry granulation method.

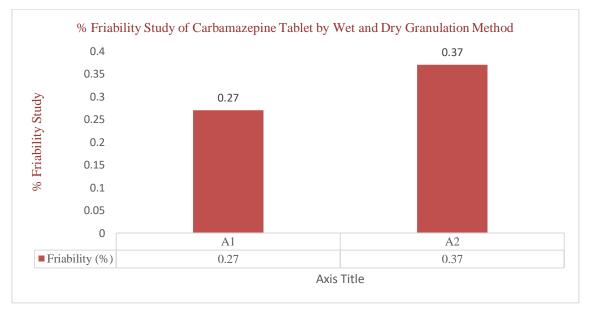
Thickness of tablets were measured and found in the range of 6.46-6.48 mm. Uniform Thickness indicates uniform die fill, good flow properties, uniform pressure and uniform punch movement. It indicates thickness of the tablet is of 0.82-0.85 mm in thickness.

d) Friability:

Tablet hardness is not an absolute indicator of strength, since some formulation compressed into very hard tablet tend to cap on attrition losing their crown portion therefore another measure of tablet strengths, its friability is often measured. The Roche friability test apparatus was used to determine the friability of the tablets. Twenty pre-weighed tablets were placed in the apparatus and operated for 100 revolutions at 25 rpm for 4 minute and then the tablets were reweighed. The percentage friability was calculated.

Formulation Code	Initial Weight of 20 Tablets	Final Weight of 20 Tablets	Friability (%)		
	(mg)	(mg)			
A1	5.069	5.055	0.27		
A2	5.112	5.093	0.37		

Table 6: % Friability Study of Carbamazepine Tablets



Graph 4: % Friability Study of Carbamazepine Tablet by Wet and Dry granulation Method

The prepared tablets showed loss in weight in the range of 0.27 to 0.37%. The percent friability was less than 1% as

ISSN No:-2456-2165

per official specification. This indicated that tablets of all formulations passed the friability test as per Pharmacopeial Standard and tablets were mechanically stable.

e) Drug Content:

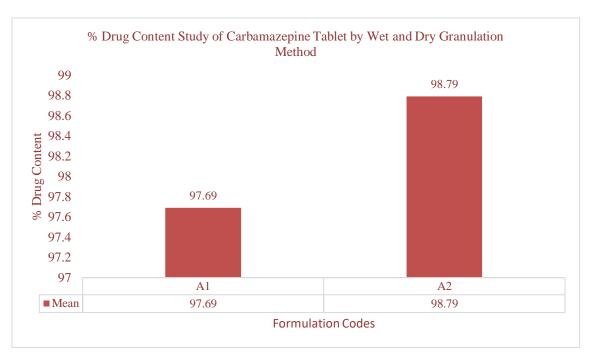
The Six tablets of 300 mg from each formulation were taken in separate 100 ml volumetric flaks

containing 1000 ml of pH 6.8 phosphate buffer and were kept for 24 hrs under constant stirring. The solutions were then filtered, diluted suitably and analyzed at 285 nm using UV spectrophotometer. The average of six tablets was taken as the content of drug in one tablet unit.

Formulation Code	1	2	3	4	5	6	Mean	SD	SEM
A1	97.65	98.32	96.92	97.11	98.44	97.72	97.69	±0.40	0.35
A2	98.75	98.42	98.85	99.22	97.65	99.85	98.79	±0.60	0.23

Table 7: % Drug Content Study of Carbamazepine Tablets

The percentage of drug content for A1 to A2 formulation was found to 97.26% to 98.97% of Carbamazepine, it complies with official specifications.



Graph 6: % Drug Content Study of Carbamazepine Tablet by wet and Dry Granulation Method

- f) Drug Release Study of Formulations of Carbamazepine Tablets:
 - I. Drug Release of A1 Carbamazepine Tablets Formulated by Wet Granulation Method:

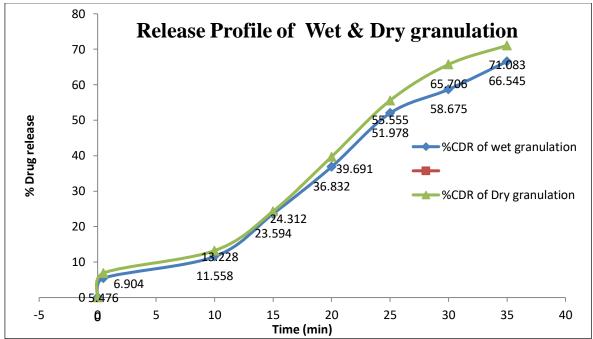
Time	Abs	Conc.	Conc.	Conc.	Conc	CDR	% CDR
(min)		(10µg/ml)	(µg/10ml)	(10 mg/10ml	10 mg/900ml		
0	0	0	0	0	0	0	0
5	0.046	1.095	10.952	0.01	5.476	5.476	5.476
10	0.097	2.308	23.095	0.023	11.547	11.558	11.558
15	0.198	4.714	47.142	0.047	23.571	23.594	23.594
20	0.309	7.357	73.571	0.073	36.785	36.832	36.832
25	0.436	10.38	103.809	0.103	51.904	51.978	51.978
30	0.492	11.714	117.142	0.117	58.571	58.675	58.675
35	0.558	13.285	132.857	0.132	66.428	66.545	66.545

Table 8: Drug Release Study of A1 Carbamazepine Tablet by Wet Granulation Method

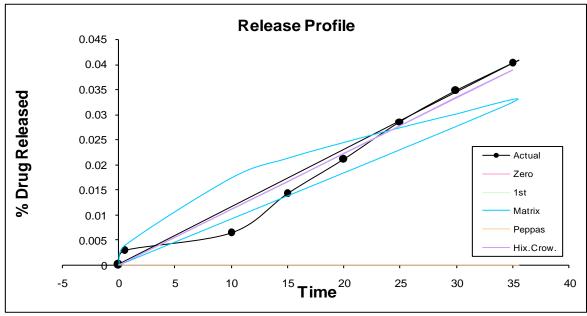
II. Drug Release of A2 Carbamazepine Tablet Formulatedby Dry Granulation Method:

Times (Min)	Abs	Conc. (10µg/ml)	Conc. (µg/10ml)	Conc. (mg/10ml)	Conc. (mg/900ml)	CDR	% CDR
0	0	0	0	0	0	0	0
0.5	0.058	1.38	13.809	0.0138	6.904	6.904	6.904
10	0.111	2.642	26.428	0.026	13.214	13.228	13.228
15	0.204	4.857	48.571	0.048	24.285	24.312	24.312
20	0.333	7.928	79.285	0.079	39.642	39.691	39.691
25	0.466	11.095	110.952	0.11	55.476	55.555	55.555
30	0.551	13.119	131.19	0.131	65.595	65.706	65.706
35	0.596	14.19	141.904	0.141	70.952	71.083	71.083

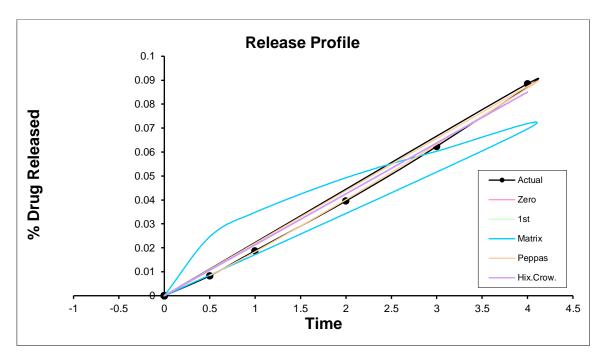
Table 9: Drug Release Study of A2 Carbamazepine Tablet by Dry Granulation Method



Graph 6: Release Profile for A1 Formulation of Carbamazepine Tablet by Dry and Wet Granulation Method



Graph 6: Model Fitting for A1 Formulation of Carbamazepine Tablet by Wet granulation Method



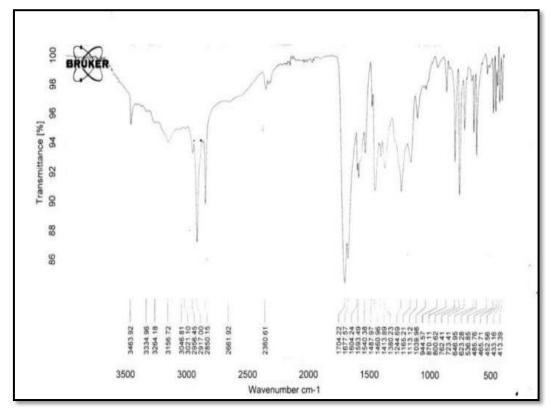
Graph 7: Model Fitting for A2 Formulation of Carbamazepine Tablet by Dry granulation Method

III. KINETIC STUDY:

Best release kinetics fitting model to all formulation represent.A1 and A2 best fitted in to zero order mathematical model which shows non fickian.

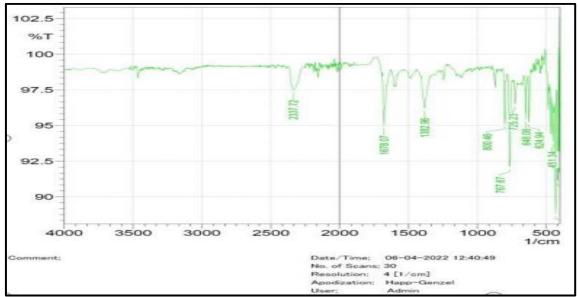
A. FTIR Study of Formulation:

• FTIR Study of Carbamazepine Tablet Formulated by wet granulation Method:



Graph 8: FTIR Spectrum of Carbamazepine Tablet by Wet Granulation Method

• FTIR Study of Carbamazepine Tablet Formulated by Dry granulation Method :

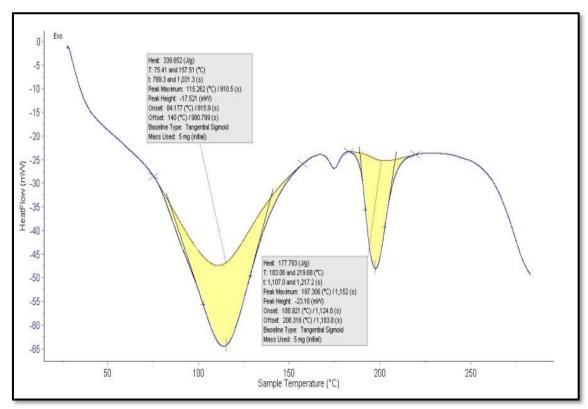


Graph 9: FTIR Spectrum of Carbamazepine Tablet by Dry Granulation Method

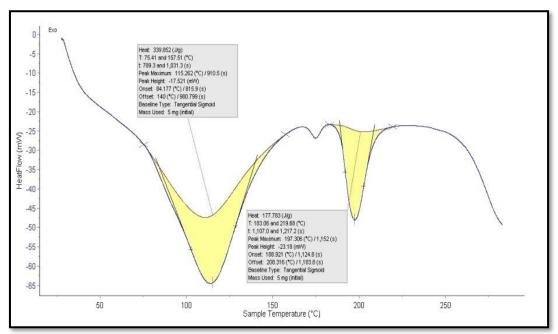
The FTIR spectrum of wet granulation method shows dominant characteristic peaks of Carbamazepine especially OH alcoholic Stretching O-H alkanes, C-H, C-O-C (ether)C=O carboxylic acid N at 3463.92, 3156.72, 2917.00, 1244.69, 1704.22, 1165.21 respectively. The FTIR spectrum of dry granulation method shows dominant characteristic peaks of Carbamazepine especially OH alcoholic Stretching O-H alkanes, C-H, C-O-C (ether) C=O carboxylic acid C-N at 2307.72, 1678.07, 1382.96, 850.46,

767.67, 725.23 respectively. The above spectrum indicated that physical changes not any chemical changes.

- B. Study of Polymorphic changes in drug Carbamazepine byusing Differential Scanning Calorimetric Analysis.
 - DSC Study of Carbamazepine in Tablet Dosage form Formulated by Using Wet and Dry Granulation Method:



Graph 10: DSC of Carbamazepine Tablet by Wet Granulation



Graph 11: DSC of Carbamazepine Tablet by Dry Granulation

IV. RESULT AND DISCUSSION

A. Differential Scanning Calorimetric Analysis of Carbamazepine in Tablet Dosage form, formulated by using Wet and Dry Granulation Method:

The polymorphic Change of Carbamazepine drug is studied for in its tablet dosage form by considering the variations in the granulation process in graph 10 and 11 (Weight and Dry). As far the DSC study recorded for the different thermogram to know the polymorphic changes happened in operating process utilizing the change in granulation technique. A sharp endothermic peak is observed at a temperature of 140°C in both granulation techniques. The drug and represented excipients melts at the specific temperature.

Then the next subsequent sharp endothermic peak is observed at 177.78°C and at 200°C for both granulation techniques is observed and indicate the drug present in its more percentage of amorphous polymorphic form. Followed by broad exothermic peak is observed from temperature range of 220°C to 280°C for both granulation technique indicates the drug present in less percentage in its crystalline polymorphic form. However the variations in granulation technique of Carbamazepine tablets never reflectpolymorphic changes even there is a change in granulation technique.

V. CONCLUSION

This study was undertaken to identify polymorphic changes in the Carbamazepine drug. Characterization of Carbamazepine tablets was carried out using Differential Scanning Calorimetry (DSC). According to DSC study the variations in granulation technique of Carbamazepine tablets never reflects a polymorphic changes even there is a change in granulation technique.

ACKNOWLEDGMENT

The Author wish to acknowledge the Head of Center of Excellence in Dairy Technology, Malegaon Khurd and Center of Innovation for Applied Science, Baramati, Dist: Pune for providing the research facility for successfully completion and execution of research work.

REFERENCES

- [1.] Bauer. S, Spanton, R. Henry, J. Dziki.W, Porter. J. Carbamazepine: an extraordinary example of conformational polymorphism, Pharmaceutical Research 2001 859.
- [2.] Brittain HG. Polymorphism: Pharmaceutical Aspect, Encyclopedia of Pharmaceutical Technology. Marcel Dekker Publication, New York 2002, 2239-2249.
- [3.] BramhankarD ,Jayswal B. Biopharmaceutics and Pharmacokinetics: A Treatise. VallabhPrakashan 2009; 5-113, 196-221.
- [4.] www.epilepsy.org.au
- [5.] Van Scoik K, Solid Pharmaceutical dosage in tablet triturates form and method of producing the same. US Patent 5,082, 667.
- [6.] Lang M, Kampf JW, Matzger AJ. Form IV of carbamazepine. J Pharm Sci. 2002; 91(4):1186-90.
- [7.] Rodryguez-Hornedo N, Murphy D. Surfactant facilitated crystallization of dihydrate carbamazepine during dissolution of anhydrous polymorph. J Pharm Sci 2004; 93(2): 449-460.
- [8.] Murphy D, Cintro.N, F, Langevin B, Kelly R, Hornedo N. Solution mediated phase transformation of anhydrous to dihydrate carbamazepine and the effect of lattice disorder. Int J Pharm 2002; 246: 121–134.
- [9.] Grzesiak L, Lang M, Kim K, Matzger J. Comparison of the four anhydrous polymorphs of carbamazepine and the crystal structure of form Int J Pharm Sci 2003; 92: 2260–2271.
- [10.] Smith R, Raftery D Analysis of Conformational polymorphism in pharmaceutical solids using solid-state NMR and electronic structure calculations.

- Journal of Physical Chemistry B 2006; 110(15): 7766-7776
- [11.] Ficarra R, Tommasini S. Solid-state study of polymorphic drugs: carbamazepine. J Pharm Biomed Anal 2000; 23: 41–54.
- [12.] Rustichelli C, Gamberini G, Ferioli V, Gamberini M Solid-state study of polymorphic drugs: carbamazepine. J Pharm Biomed Anal 2000; 23: 41– 54.
- [13.] Padmalata H, Development and Validation of UV-Spectrophotometer for the estimation of Erlotinib in tablet formulation. Imp J Med Org Chem.2011; 1(1):28-30.
- [14.] Indian Pharmacopoeia, Government of India, ministry of health and familywalfare, Ghaziabad. The Indian Pharmacopoeia commission. 1996; (2):554-556.
- [15.] Robert MS, Fransis XW, Spectrometric Identification of Organic Compounds. Wiley010; (6):79-109.
- [16.] Gurdeep RC, Anand SK, Instrumental Methods of Chemical Analysis. Mumbai. Himalaya Publishing House. 2010; (5):2.747-2.752.
- [17.] US Pharmacopoeia 30 NF 25. 2007;1225-1227.
- [18.] Silverstein RM, Webster FX, Spectometric Identification of Organic Compounds. Wiley 2010; 6: 79-109.
- [19.] JavadzadehY, Jafari B, Nokhodchi A. Liquisolid Technique for Dissolution Rate Enhancement of A High Dose Water-Insoluble Drug (Carbamazepine) Int J Pharma 2007; 341: 26–34.
- [20.] MJ, Fites AL, Robinson RL. Types of parenteral administration. Journal of parenteral science and Technology, 1987; 41: 88-95.
- [21.] Lippincolt, Williams K. Remington, The Science & practice of pharmacy, Parenteral Preparation, 20th ed, ISE publication, Phelabelphia. 2000; 1: 804-819.
- [22.] Akers MJ, Remington: The science and practice of pharmacy, Lippincott Williams &wilkins publisher, 2000; 21: 525.
- [23.] Chien&Yiew W. Pharmaceutical Dosage forms: Parenteral Medications. Indian Journal of pharmaceutical science and technology, 1981; 35: 106-118.
- [24.] Liberman HA, Lachman L and Schwartz BJ. Pharmaceutical dosage form: Parenterals, Marcel Dekker publisher, 1989.
- [25.] Neema S, Washkuhn RJ and Brendel RJ. Injectable products. PDA J Pharm SciTechnol, 1997; 51: 166-171.
- [26.] Gandhi S and Chandrul K. Pharmaceutical Solid Polymorphism in Abbreviated New Drug Application (ANDA) - A Regulatory Perspective. J Chem Pharm Res 2011; 3(3):6-17.
- [27.] Omar M, Makary P and Wlodarski M. A Review of Polymorphism and the Amorphous State in the Formulation Strategy of Medicines and Marketed Drugs. UK Journal of Pharmaceutical and Biosciences 2015; 3(6):60-6.
- [28.] Raza K, Kumar P, Ratan S, et al. Polymorphism: The Phenomenon Affecting the Performance of Drugs. SOJ Pharm PharmSci 2014; 1(Pharm Res 2011; 3(3).

- [29.] Wang W. Lyophilization and development of solid protein pharmaceuticals. Int J of pharmaceutics,2000; 52: 1-60.
- [30.] Jennings TA. Effect of formulation on lyophilization. Asian journal of pharmaceutical science, 1997; 54-63.
- [31.] Inamdar N, Rathi A A and Saifee M. Drug Polymorphism: A Review. International Journal of Health Research 2009; 2(4):291-306.
- [32.] Yoshioka S, Aso Y and Kojima S. The effect of excipients on the molecular mobility of lyopihilized formulations, as measured by glass transition temperature and NMR relaxationbased critical mobility temperature. Pharm Res, 1999; 135-140.
- [33.] Sanjith NL &Gatin LA. Freeze drying: Annealing principles and practice. NP publication. 1993; 2: 163-233.
- [34.] Gatin LA, Auffret T, Shalaev EY, Speaker SM and Teagarden DL. Freeze Drying Concepts: The Basics in Formulation and delivery, Informa Healthcare, New York, 2008; 15: 177-195.
- [35.] Greiff D. Development of cycles for lyophilization. DevBiol Stand, 1992; 74: 85-92.
- [36.] Carpenter JF, Pikal MJ, Chang BS and Randolph TW. Rational design of stable lyophilized protein formulations: some practical advice. Pharm Res, 1997; 14: 969-975.
- [37.] Kadam K P and Chavan R P. Evaluation of Various Polymorphs by Different Techniques and Their Characterization A Review. Int J of Engineering and Science 2016; 5(6):29-34.
- [38.] S.J. Dengale, H. Grohganz, T. Rades, K. Löbmann, Recent advances in co-amorphous drug formulations, Advanced Drug Delivery Reviews, 100 (2016) 116-125.
- [39.] K. Löbmann, H. Grohganz, R. Laitinen, C. Strachan, T. Rades, Amino acids as co-amorphous stabilizers for poorly water soluble drugs—Part 1: Preparation.