

# Phytochemical Screening, Physicochemical and Antioxidant Activity, TLC & Finger Print of HPTLC, *Morus Alba* Ethanol Extraction

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**Abstract:-** The present study is design to evaluate preliminary phytochemical constituents, physiochemical properties evaluation, TLC & HPTLC fingerprinting analysis of *Morus alba* Linn (White mulberry) Leaves powder extraction.

**Total phenol content, total flavonoid content, Total tannin content, TLC, HPTLC fingerprinting analysis for compounds responsible for antioxidant activity.**

Phytochemical screening shows the presence of constituents like alkaloids, glycosides, carbohydrates & phytosterols, saponin, phenolic compounds, tannins, proteins, amino acids, gums and mucilage. Physicochemical studies shows that the powder was contain a moisture content 12.5% and ash value 7.5%. Total phenol content was 3.44 mg gallic acid equivalent/g, total flavonoid content was 5.00 mg rutin equivalent/g and total tannin content was 3.00 mg tannic acid equivalent/g. HPTLC fingerprinting also proofs the presence of phenol & flavonoid compounds in the leaf powder.

The results in the paper shows that the *Morus alba* Linn (White mulberry) leaf powder extract is an natural antioxidant and can be used for treatment of various disorders.

## I. INTRODUCTION

Natural plants are used as very good source of nutrition persistent food as well as source of various chemical constituents operative in curing various diseases which may demand as the biologically active constituents. At the present natural plants are very much in petition in the form of drugs because of their fewer side effects, they are considered the potential resources of various bioactive compounds and are also easily available from the natural sources. In the same context *Morus alba*, the Mulberry plant which is basically famous for sericulture, the fabrication of silk done through the silkworm and the leaves are also used to diminish the symptoms of diabetes in vernacular medicine as well as for improving cardio-metabolic risks, including antihyperglycaemic, antihyperlipidaemic, antiobesity, antihypertensive, antioxidative, anti-inflammatory, anti-atherosclerotic and cardioprotective effects<sup>1</sup> in Chinese medicine used to treat constipation, to tonify the blood,

prematurely grey hair, cough, edema, to promote urination, fever, headache, dry & sore eyes<sup>2</sup> and so many more, So, the leaves is used further in this study to explore some more about the biological activity of leaves.

Mulberry plant belongs to genus *Morus* having 68 species which are unisex flowering plants belonging to family Moraceae of the Utricles subclass. The plant is a shrub or tree (20 to 30 feet high) often the size of a small apple tree, having leaves which are thin, glossy, and light green in color with 5 lobes or has one lobe, two lobes, three lobes, or no lobes at all. *Morus Alba* L is also known as White Mulberry and can be grown from seed as well as planted from large cuttings of root readily. Commonly, the plantation is upraised in a block foundation with arrangement of 6 feet x 6 feet, or 8 feet x 8 feet, as plant to plant and row to row spaces. The plants are generally trimmed once a year during the monsoon season (July – August) to a height of 5–6 feet and allowed to grow with a maximum of 8–10 shoots at the top.<sup>3</sup> The plant is widely distributed in India, China, Japan, North Africa, Arabia, South Europe, etc.

*Morus alba* L. leaves had been used as substantial source of medicine, drink, and functional foods in many countries. It is used in drinks as green tea with several other herbal drugs like tulsi and aswagandha because of its immune boosting antioxidants like Chlorogenic acid, rutin, isoquercitrin, and astragaline. Anticancerous alkaloids like 1-deoxynojirimycin, morrols B–F<sup>4</sup>, (2R,3R,4R)-2-hydroxymethyl-3,4-dihydropyrrolidine-N-propionamide from the root bark and 4-O-R-D-galactopyranosyl-calystegine B2 and 3β,6β-dihydroxynortropane from the fruits<sup>5</sup>, mulbaines A, B & C<sup>6</sup>. Eighteen important amino acids with calcium, potassium, sodium, magnesium, zinc, iron, copper, manganese, chromium, selenium, arsenic, vitamins and it's no caffeine property. Other chemical constituents present in leaves are coumarins, flavonoids, anthocyanins and polyphenols including quercetin 3-(malonylglucoside), rutin, isoquercitrin, cyaniding-3-rutinoside apigenin, luteolin, quercetin, morin, caffeic acid, gallic acid, umbelliferone, chlorogenic acid, and kaempferol.<sup>7</sup> The plant extract rich in polyphenols used as a non-toxic natural healing agent, which also have high prospective applications as skin-whitening agents due to its potent tyrosinase inhibitor property.<sup>8</sup>



Fig 1-Morus Alba Linn plant.

➤ *Classification of Morus alba Linn*

Kingdom	Plantae – Plant
Subkingdom	Tracheobionta-Vascular Plant
Super division	Spermatophyta
Division	Magnoliophyta – Flowering Plant
Class	Magnolipsida – Dictyledons
Subclass	Hamamelididae
Order	Urticales
Family	<i>Moraceae</i>
Genus	<i>Morus L.</i> (Mulberry)
Species	<i>Morus alba, Morus nigra, Morus rubra, Morus indica, Morus serrate, Morus mongolica.</i> <sup>6</sup>

➤ *Same species of Moraceae family with their pharmacological uses:*

S.No.	Plants Species	Parts	Uses
1	<i>Morus mesozygia</i>	Leaves and fruits	Arthritis , Wound healing, Antioxidants and Antimicrobial compounds.
2	<i>Morus nigra</i>	Fruits	Anti-diabetic, Anti-oxidative, Anti-hyperlipidemia, Anti-inflammatory.
3	<i>Morus indica</i>	Root	Anti-inflammatory activities.
4	<i>Morus alba L.</i>	Fruit	Antioxidant, Neuroprotective, Antitumor, Immunomodulative activities. <sup>7</sup>

➤ *Vernacular names-*

<b>English</b>	White mulberry
<b>Hindi</b>	Sahtoot, Shahtoot, Shahtut, Shehtoot, Swa, Toot, Toota, Tooth, Tul, Tulklu, Tunt, Tut, Tut.
<b>Sanskrit</b>	Tooda, Toola, Tudah, Tula, Tutam.
<b>Tamil</b>	Kamblichedi, Kampalliceti 1, Kampilippucciceti, Musukotta, Pattuppucci, Pattuppuccimaram.
<b>Kannada</b>	Bilee hippenerale, Bili uppu naerale, Hippal verali, Hippali naerale, Hippu naerale, Hippunerale, Kambali gida, Korigida, Tuti, Uppunute. <sup>8</sup>

➤ *Description of Plant*

*Morus alba Linn* (white mulberry) plant has medium leaves (8-15 cm.), edges or margins. The leaves are simple, alternate and shallowly to deeply lobed. The flowers will be white & light yellow and the fruit is white and sweet taste. Seeds are very small in size & medicinal used. The white mulberry plant has been grown a normal temperate. The mulberry cultivate the fruit in April month. The stem is brown color and stem used in a medicinal and the family of *Moraceae*.<sup>9</sup>



Fig 2- Leaf veins and venation patterns reticulate, smaller veins lording and network.

➤ *Macroscopic Characteristics-*

Leaves		Fruit	
Colour	pale green	Colour	white
Order	characteristics	Order	sweetness
Size	8-15 c.m.	Size	2-4 c.m.
Taste	acid	Taste	sweet <sup>10</sup>

➤ *Chemical Constituents of Morus alba Linn-*

*Morus alba Linn.* (white mulberry plants) is medicinal plant it contains following chemical constituents :( leaves, roots, frites, flower, stem.) Albino, Chalcomoracin, Moracin-c, Morin, Mulbarrofurana Q, Butain, Marrola A, 2-formyl-1H-pyrrole-1-butanol acid, Gallic acid, Tannic acid, Chlorogenic acid, Benzoic acid, Rutin astragaline, beta-sitosterol-3-O-beta-D-glucoside, beta-sitosterol, Salvigenin, Cirsimaritin, Quereetin.<sup>7,10</sup>

➤ *Bioactive Substances Occurring in White Mulberry*

S.No.	Plant parts	Substances	Active compounds
1	Leaves	Fatty acids	Palmitic acid, Linoleic acid, Eicosanoids, Oleic acid,
		Phenolic compounds	Kaempferol
2	Fruits	Phenolic compounds	Quercetin, Morin, Chlorogenic acid, Hydroxybenzoic acid, Ferulic Acid, Rutin,
		Fatty acids	Palmitic acid, Oleic acid, Linoleic acid
		Glucosides	Rutinoside cyanine, cyanine glucoside
		Phenolic compounds	Maklurein, Rutin, Isoquercetin, Resveratrol, Morin, Apigenin,
3	Stem bark	Triterpenoids	Moruslupenoic acid A, Moruslupenoic acid B, Moruslanosteryl acetate
		Mulberry flavonoids	Sanggenols, Kwanon, Mulberrofuran
4	Root bark	Phenolic compounds	Luteolin gallic acid, Sinapic acid, Resveratrol
		Lectins	Albianol <sup>11</sup>



**II. MATERIAL METHOD-**

**Table: Morphological Study**

S.No	Parameters	Characteristics features
1	Colour	Yellowish
2	Order	Characteristics
3	Taste	Sweet
4	Nature	Coarse powder



Fig 3: Morus Alba Fruits



Fig 4: Barks



Fig 5: Leaves powder



Fig 6: Leaves

➤ **Soxhlet Extraction Procedures**

*Morus alba* leaves was collected in Aryakul campus then fresh leaf wash and dry at room temperature. All leaves dry (after two weeks) & crushed with the help of in Mortars & Pestles. The powdered material used for extracted by Soxhlet apparatus using

the solvents for Petroleum ether, Chloroform, Acetone, Ethanol and Methanol successively. The extraction process a (40-60<sup>0</sup>C) temperature for 6 hours. The concentrated product was collected and stored in refrigerator for further experimental analysis.<sup>16,17,18</sup>

➤ **Preliminary Phytochemical Screening**

The Petroleum ether, Chloroform, Acetone, Ethanol and Methanol extracts of *Morus alba Linn.* (White mulberry) will be screened for following phytoconstituents class. The method for the preliminary phytochemical screening was carried away as per, Pharmacognocny.<sup>10</sup>

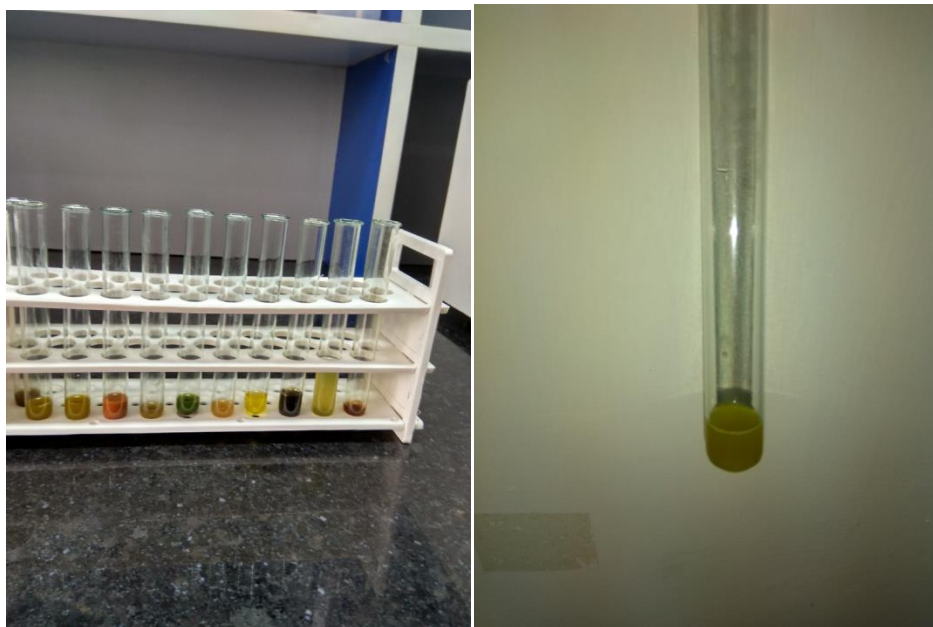


Fig 7: Chemical test

➤ *Thin Layer Chromatography-*

The TLC plates were prepared by using silica gel G, and were left for air drying. These Plates were activated by hot air drying in hot air oven at (80-100<sup>0</sup>C) for 1 hr. Extracts from different solvents was spotted on the TLC plates. The plates were dried and developed in suitable solvents for rapid screening. The plates was run in the following solvent system and dried at room temperature (25<sup>0</sup>C). The detection of TLC plate was done by Iodine chamber and UV chamber<sup>21</sup>. R<sub>f</sub> value of different spots available is calculated by using the formula-

$$R_f \text{ value} = \text{Distance travelled by the solute} / \text{Distance travelled by the solvent}$$

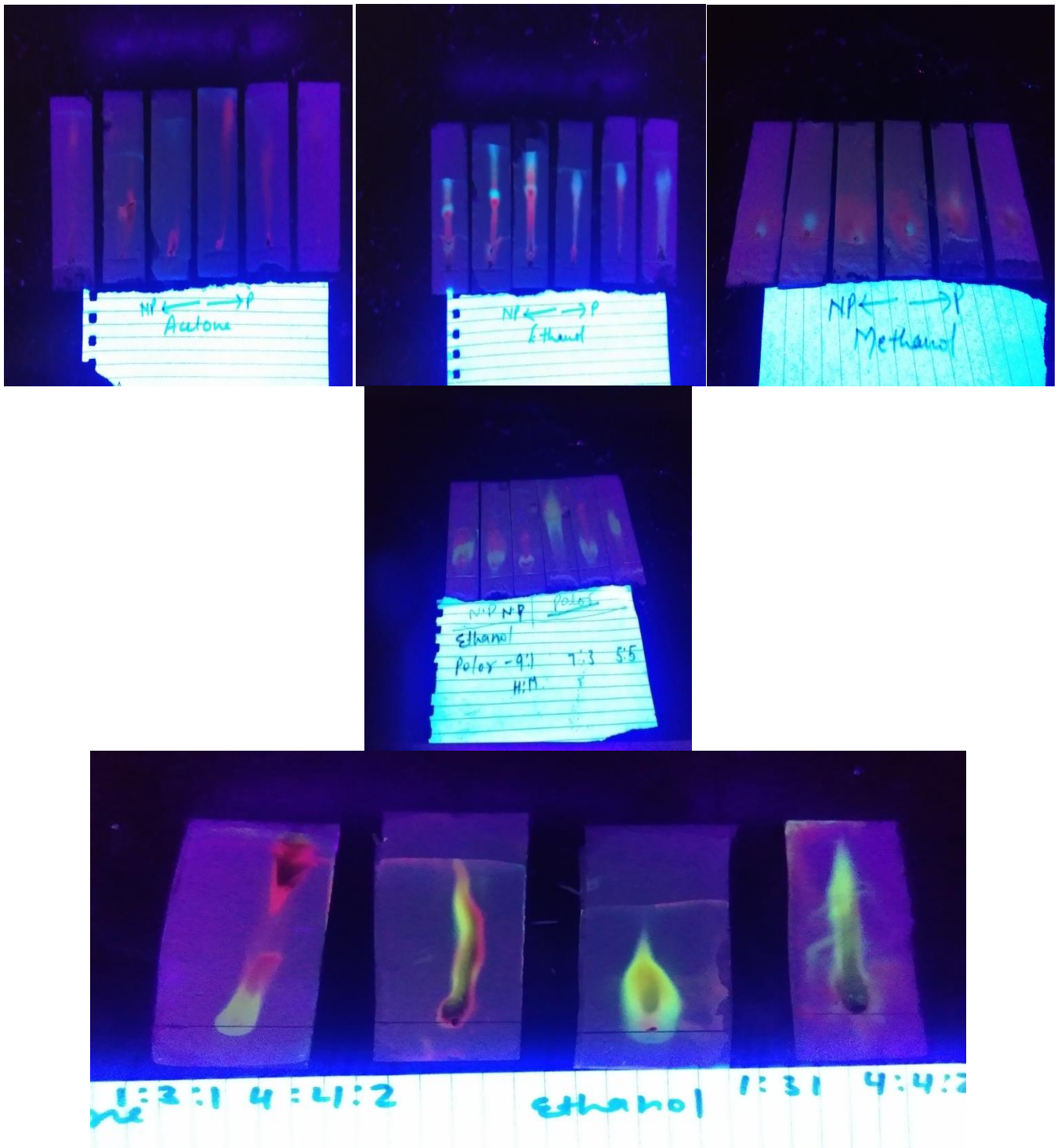


Fig 8: Visible under the U.V. chamber  
Detector: UV Chamber



Fig. 9:- TLC performance in Laboratory

➤ *Development of HPTLC Method for Analysis of Flavonoids*

HPTLC analysis was carried out using Camag HPTLC system equipped with Linomat-v applicator and 100 ul syringes. The samples were spotted in the form of bands using micro liter syring on pre-coated silica gel 60 F254 HPTLC plates and development of the applies plate was carried out in pre-saturated Camag twin-trough chamber<sup>25</sup>. The developed plates were dried and analysed at 254nm and 366nm. The mobile phased used flavonoids is ethyl acetate: acetic acid: Formic acid: Water (10:1.1:1.1:2.3)<sup>11</sup>

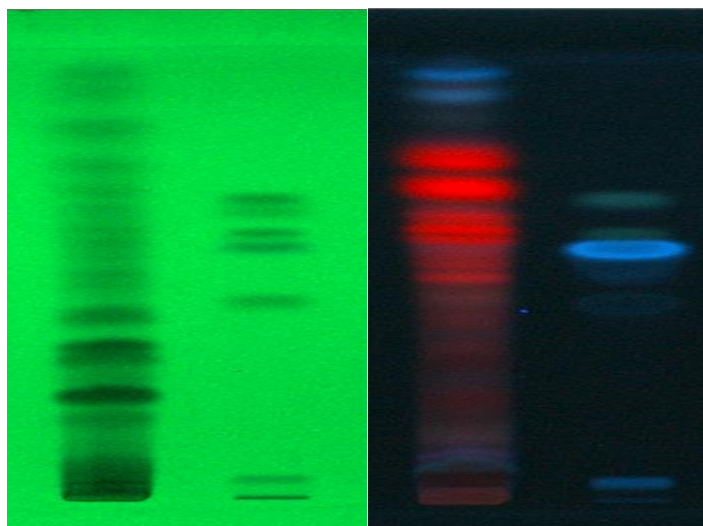


Fig. 10:- High - performance thin layer chromatography plate

### III. ANTIOXIDANT CONTENT

#### *A-Estimation of Total Phenolic Content*

Phenolics were determined using gallic acid as standard. A 100µg/ml stock solution of gallic acid was prepared. From the above stock a 0.5ml aliquot was pipette out into 25ml volumetric flask. 10ml distilled water and 1.5ml Folin ciocalteus reagent was added. Then after 10 min

4 ml 20% sodium carbonate was added and volume is makeup up to 25ml using distilled water.<sup>22</sup>

A stock solution of 1mg/ml in methanol of ethanolic extract was prepared. From the above stock 0.5ml of extract was taken in 25ml volumetric flask. 10ml distilled water and 1.5ml Folin ciocalteus reagent was added. Then after 10 min 4 ml 20% sodium carbonate was added and volume is makeup up to 25ml using distilled water.

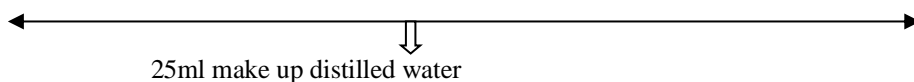


After 30min the absorbance of both test and standard solution was taken at 765nm. Percentage of total phenolics was calculated using the equation based on calibration curve of gallic acid:

$$Y = 0.198x - 0.031$$

$$R^2 = 0.994$$

Chemical	Standard	Standard	Standard	Standard	Standard	B	Ethanol sample	Methanol sample
	1	2	3	4	5			
Gallic acid (10%)	0.2ml	0.4ml	0.6ml	0.8ml	1ml	---	0.5ml	0.5ml
Distilled water	10ml	10ml	10ml	10ml	10ml	10ml	10ml	10ml
Folin	1.5ml	1.5ml	1.5ml	1.5ml	1.5ml	1.5ml	1.5ml	1.5ml
Na <sub>2</sub> CO <sub>3</sub>	4ml	4ml	4ml	4ml	4ml	4ml	4ml	4ml



**B-Estimation of Total Flavonoid Content**

Flavonoids were determined using rutin as standard. A 100µg/ml stock solution of rutin was prepared. From the above stock a 0.5ml aliquot was pipette out. 0.5ml 2% metha) methanolic aluminium chloride, few drops of distilled water and 4ml methanol was added.

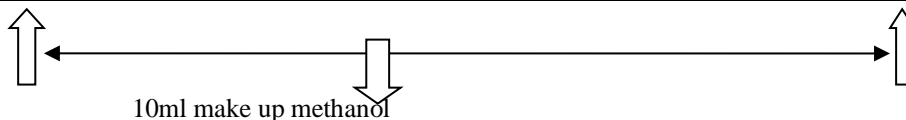
A stock solution of 1mg/ml in methanol of ethanolic extract was prepared. From the above stock 0.5ml of extract was taken, add 0.5ml 2% ethanolic aluminium chloride, few drops of distilled water and 4ml methanol was added.

After 20min the absorbance of both standard and test solution was taken at 420nm. Percentage of total flavonoids was calculated using the equation based on calibration curve of rutin.<sup>23</sup>

$$Y = 0.016x - 0.204$$

$$R^2 = 0.926$$

Chemical	Standard	Standard	Standard	Standard	Standard	B	Ethanol sample	Methanol sample
	1	2	3	4	5			
Rutin	0.2ml	0.4ml	0.6ml	0.8ml	1.0ml	---	0.5ml	0.5ml
AlCl <sub>3</sub>	0.5ml	0.5ml	0.5ml	0.5ml	0.5ml	0.5ml	0.5ml	0.5ml
Methanol	4.3ml	4.3ml	4.3ml	4.3ml	4.3ml	4.3ml	4.3ml	4.3ml



**C-Estimation of Total Tannin Content**

Tannins were determined using tannic acid as standard. A 100µg/ml stock solution of tannic acid was prepared. From the above stock a 1ml aliquot was pipette out in 100ml volumetric flask, 5ml Folin ciocalteus reagent, few ml distilled water, 10ml saturated sodium carbonate solution were added and volume was make up by distilled water up to 100ml.

2gm powdered material were extracted and 1ml aliquot was pipette out in 100ml volumetric flask, 5ml Folin ciocalteus reagent, few ml distilled water, 10ml saturated sodium carbonate solution were added and volume was make up by distilled water up to 100ml.

Absorbance of both standard and test solution was taken at 760nm. Percentage of total tannin was calculated using the equation based on calibration curve of tannic acid.<sup>24</sup>



$$Y = 0.043x - 0.053$$

$$R^2 = 0.0994$$

Chemical	Standard	Standard	Standard	Standard	Standard	B	Leaf + water extract	Leaf + water Extract
	1	2	3	4	5	6	7	8
Tannic acid	1ml	2ml	3ml	4ml	5ml	----	1ml	1ml
Folin	5ml	5ml	5ml	5ml	5ml	5ml	5ml	5ml
H <sub>2</sub> O	2ml	2ml	2ml	2ml	2ml	2ml	2ml	2ml
Na <sub>2</sub> CO <sub>3</sub>	10ml	10ml	10ml	10ml	10ml	10ml	10ml	10ml

➤ *Chlorophyll Estimation (Leaf powder, Morus alba linn )*

S.No.	Wave length	Absorbance
1	645	0.510
2	663	1.177
3	652	0.657

Chlorophyll A Content	Chlorophyll B Content	Chlorophyll Total
1.357	0.647	1.974

#### IV. RESULT AND DISCUSSION

➤ *ORGANOLAPTIC STUDY*

**Table- Organoleptic study**

S.NO.	Parameters	Characteristics features
1	Color	Green
2	Order	Characteristics
3	Taste	Acrid
4	Nature	Coarse powder

**Table-Extract Characteristics**

Types of solvent	Consistency	Color	Extractive value% w/w
Petroleum ether	Oily	Dark green	2.75%
Chloroform	Oily	Light yellow	140.85%
Acetone	Pasty	Green	17.05%
Ethanol	Pasty	Dark green	36.60%
Methanol	Pasty	Dark green	63.70%

**Table-Phyto-chemical screening**

(A) *ALKALOIDS-*

S.No.	Plant constituents test/reagent used	Petroleum ether	Chloroform	Acetone	Ethanol	Methanol
1	Meyer's reagent	+	+	+	+	+
2	Dragendroff's reagent	+	+	+	+	+
3	Wagner's reagent	+	+	+	+	+
4	Hager's reagent	-	-	+	+	+

(B) *TEST FOR CARBOHYDRETES-*

Extract	Pt. ether	Chloroform	Acetone	Ethanol	Methanol
Molisch's Test	–	+	+	+	+
Barfoed's Test	–	–	–	–	–
Test for pentoses	+	–	+	+	+

*(C) TEST FOR AMINO ACIDS-*

Extract	Pt. ether	Chloroform	Acetone	Ethanol	Methanol
Millon's	+	–	+	+	+
Ninhydrine	–	–	–	–	–

*(D) TEST FOR FLAVONOIDS-*

Extract	Pt. ether	Chloroform	Acetone	Ethanol	Methanol
Alkaline reagent test	–	–	–	+	+
Zinc hydrochloride test	–	+	+	–	–

*(E) TEST FOR ANTHRAQUINONE GLYCOSIDES-*

Extract	Pt. ether	Chloroform	Acetone	Ethanol	Methanol
Test for hydroxyl	+	–	–	+	+
Bronteager's test	–	–	–	+	+

*(F) TEST FOR CARDIAC GLYCOSIDES-*

Extract	Pt. ether	Chloroform	Acetone	Ethanol	Methanol
Baljets test	–	–	+	+	+
Raymond's test	–	–	–	+	–
Legal's	–	–	–	+	+

*(G) TEST FOR SAPONIN GLYCOSIDES-*

Extract	Pt. ether	Chloroform	Acetone	Ethanol	Methanol
Forth formation test	+	–	–	–	+
Haemolysis test	–	–	–	–	–

*(H) TEST FOR TANNINS (PHENOL COMPOUNDS)-*

Extract	Pt. ether	Chloroform	Acetone	Ethanol	Methanol
Feric chloride test	–	–	–	–	–

*(I) TEST FOR STEROIDES & TRI TERPENOIDES-*

Extract	Pt. ether	Chloroform	Acetone	Ethanol	Methanol
Salkowski test	–	–	–	+	–
Sulfur powder test	–	–	–	–	–

*(J) TEST FOR NAPHTHOQUINONES-*

Extract	Pt. ether	Chloroform	Acetone	Ethanol	Methanol
Dam Karrer test	–	–	+	–	–

*(K) TEST FOR STARCH-*

Extract	Pt. ether	Chloroform	Acetone	Ethanol	Methanol
Starch (Amylum)	–	–	–	+	–

**Table-TLC Studies (Calculation of R<sub>f</sub> Value)-**

Extract	Non Polar						Polar						
	Mobile Phase A		Mobile Phase B		Mobile Phase C		Mobile Phase D		Mobile Phase E		Mobile Phase F		
	Spot no.	Rf value	Spot no.	Rf value	Spot no.	Rf value	Spot no.	Rf value	Spot no.	Rf value	Spot no.	Rf value	
Pt. ether	2	0.62	1	0.47	1	0.62	2	0.36	2	0.83	1	0.61	
		0.44						0.22		0.51			
Chloroform	1	0.52	1	0.33	1	0.21	1	0.52	2	0.48	1	0.84	
										0.33			
Acetone	1	0.91	2	0.92	1	0.22	1	0.91	1	0.52	1	0.91	
				0.59									
Ethanol	2	0.72	2	0.88	3	0.96	1	0.94	1	0.95	1	0.82	
													0.61
													0.43
Methanol	1	0.64	1	0.32	1	0.35	1	0.44	1	0.42	1	0.59	

**(L) TEST FOR PROTEINS-**

Extract	Pt. ether	Chloroform	Acetone	Ethanol	Methanol
Hydrolysis test	-	-	-	-	-

**(M) TEST FOR FATS & FIXED OILS-**

Extract	Pt. ether	Chloroform	Acetone	Ethanol	Methanol
	-	-	-	-	-

+ mean present, -mean absent

**Table-Physico - chemical Parameter**

S. No	Physico - chemical Parameter	Value
1	Moister value	12.5%
2	Ash value	7.5%

**V. ANTIOXIDANT CONTENT****Table-Total Phenolics, Flavonoids & Tannins content of Ethanolic extracts-**

Plant extracts	Total phenolics (mg gallic acid equivalent/g)*	Total flavonoid (mg rutin equivalent/g)*	Total tannin (mg tannic acid equivalent/g)*
Ethanolic extract	3.44	5.00	3.00

**➤ Chlorophyll Estimation (Leaf powder, Morus alba linn)**

S.No.	Wave length	Absorbance
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Chlorophyll A Content	Chlorophyll B Content	Chlorophyll Total
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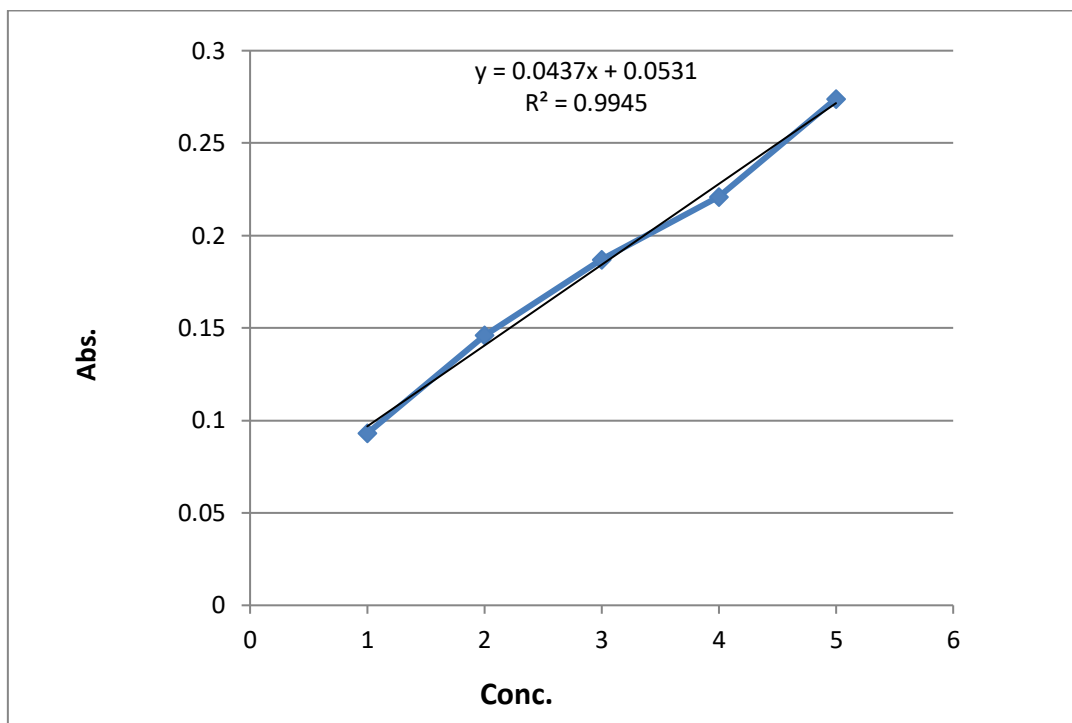


Fig 11: Standard curve of Tannin content.

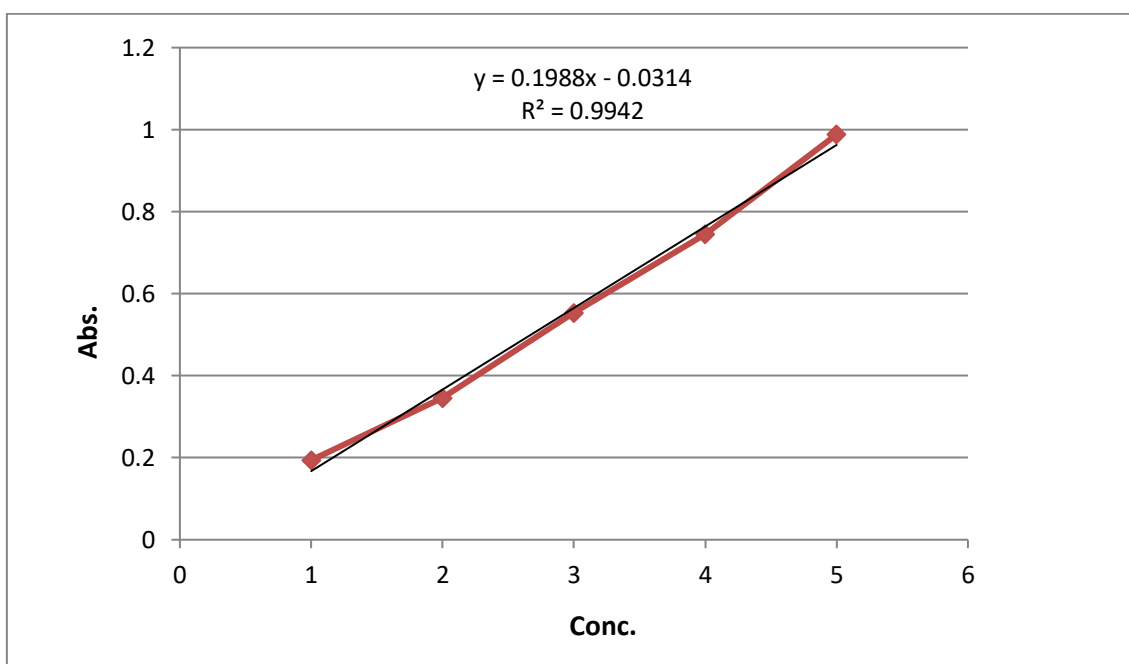


Fig 12: Standard curve of Phenol content



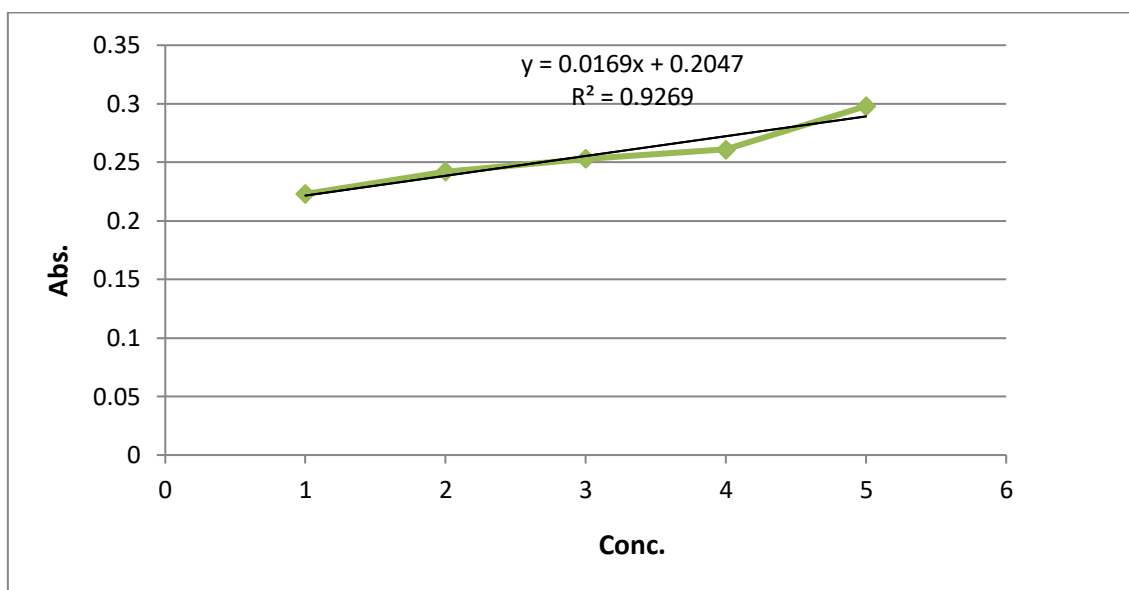


Fig 13: Standard curve of Flavonoid content

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